

**THE STRUCTURE AND STEREOCHEMISTRY OF  
BARRIGENIC ACID, A NEW TRITERPENE ACID  
SAPOGENIN FROM BARRINGTONIA ACUTANGULA**

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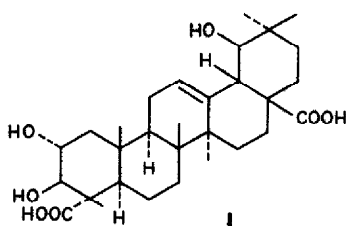
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**Key Word Index**—*Barringtonia acutangula*; Barringtoniaceae; new triterpene acid; barrigenic acid; 2 $\alpha$ ,3 $\beta$ ,19 $\beta$ -trihydroxyolean-12-en-23,28-dioic acid.

**Abstract**—A new triterpene acid, barrigenic acid, was isolated from the fruits of *Barringtonia acutangula*. Its structure was established as 2 $\alpha$ ,3 $\beta$ ,19 $\beta$ -trihydroxyolean-12-en-23,28-dioic acid.

From the branch wood of *Barringtonia acutangula* the isolation of a new triterpene acid sapogenin, called barrigenic acid, was reported previously from this laboratory [1]. This paper reports the isolation of its 19 $\beta$ -isomer, barrigenic acid (1), from the fruits of the plant.



Barrigenic acid,  $C_{30}H_{46}O_7$ , mp 286–90° (dec),  $[\alpha]_D^{25} + 80.2^\circ$  (Py) furnished a dimethyl ester,  $C_{32}H_{50}O_7$ , mp, 252–54°,  $[\alpha]_D^{33} + 75^\circ$  ( $CHCl_3$ ), with ethereal diazomethane, which gave a pale yellow colouration with tetranitromethane (UV 205 nm ( $\log \epsilon$  3.43), trisubstituted double bond; IR,  $\nu_{max}$   $cm^{-1}$ : 3420 (–OH), 1730 and 1700 (–COOMe). Dimethyl barrigenate treated with  $Ac_2O$ -pyridine at 100° gave an amorphous triacetate,  $C_{38}H_{56}O_{10}$ , mp, 145–50°, and a crystalline diacetate,  $C_{36}H_{54}O_9$ , mp 212–14°,  $[\alpha]_D^{27} + 12.5^\circ$  ( $CHCl_3$ ), the latter being the major product. Dimethyl barrigenate consumed nearly 1 mole of periodic acid indicating an  $\alpha$ -glycol system. Oxidation of the diacetate with  $CrO_3$  in acetic acid furnished a ketone,  $C_{36}H_{52}O_9$  ( $M^+$  628), mp 247–49°,  $[\alpha]_D^{27} + 55.3^\circ$  ( $CHCl_3$ ) (UV showed this not to be an  $\alpha$ : $\beta$ -unsaturated ketone), which on treatment with ethanolic caustic potash gave a conjugated ketone,  $C_{32}H_{48}O_7$ , mp 192–94°,  $[\alpha]_D^{27} - 169.3^\circ$  ( $CHCl_3$ ); UV 249 nm. This showed that the hindered OH group was at C-19 and hence the ketone diacetate was identical

with that formed from diacetyl dimethyl barrigenate. Thus the new compound was probably the 19 $\beta$ -isomer. This was confirmed by the following experiments. The diacetate, on treatment with  $POCl_3$  in pyridine, gave an unresolvable mixture, which showed no absorption maxima (UV) for either a  $\Delta^{11,13(18)}$ -diene or a  $\Delta^{12,18}$ -diene system. The mixture was saponified with alcoholic KOH and the NMR spectrum of the product showed signals at  $\delta$  6.63 and 4.77 for exomethylene protons ( $>C=CH_2$ ) and also another signal at  $\delta$  1.68 for a double bond attached methyl group ( $>C=C-CH_3$ ) suggesting that dehydration led to the expected products formed by a Wagner–Meerwein type rearrangement [2–4].

The above result suggested that the 19–OH group in barrigenic acid is  $\beta$ . The MS of both dimethyl barrigenate and its diacetate showed no molecular ion peak at  $m/e$  546 or 630 but a peak at  $m/e$  528 or 612 due to the facile loss of water. Other peaks in the MS of both compounds showed loss of COOMe from C-4 and  $m/e$  278 equivalent to the C/D ring system.

The NMR spectrum ( $CDCl_3$ , 60 MHz) of diacetyl dimethyl barrigenate showed six quarternary methyl groups ( $\equiv C-CH_3$ ) as sharp singlets at  $\delta$  0.75, 0.87, 0.91, 1.08, 1.17 and 1.27. The high field shift of one of the methyl signals to  $\delta$  0.75 showed the presence of a carbomethoxyl group at C-17. The two vicinal acetoxy groups ( $-O-COCH_3$ ) at C-2 and C-3 gave singlets at  $\delta$  1.98 and 2.07; the singlets at  $\delta$  3.63 (3H) and at 3.7 (3H), being due to the C-28 and C-23  $COOCH_3$  groups respectively. The presence of a C-23 carbomethoxyl group caused a downfield shift of the signal for the 24-methyl group [5] to  $\delta$  1.27. The 3-axial ( $\alpha$ ) proton appeared as a doublet at  $\delta$  4.84 ( $J$  10.5 Hz). The large coupling constant showed the *trans* (diaxial) relationship of this proton with the 2 $\beta$ -proton which appeared as a broad multiplet centred at  $\delta$  5.7.

The 19-axial ( $\alpha$ ) proton appeared as a doublet at  $\delta$  3.39 ( $J$  11 Hz) by coupling with the 18 $\beta$ -H (2.58,  $d$ ,  $J$  11 Hz) which are in a *trans* diaxial relationship. The C-12 vinyl proton appeared as a triplet at  $\delta$  5.44 ( $J$  3.5 Hz). Thus the NMR spectral data is in good agreement with

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the structure **1** for 2,3-diacetyl dimethyl barrigenate. The NMR spectrum of the triacetate was also in agreement.

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### THE OCCURRENCE OF 24-METHYLENELANOST-8-EN-3 $\beta$ -OL IN THE SEEDS OF *BRASSICA NAPUS*

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**Key Word Index**—*Brassica napus*; Cruciferae; triterpene alcohol; 24-methylenelanost-8-en-3 $\beta$ -ol; seeds.

**Plant and source.** The seeds of *Brassica napus* were courteously supplied by Dr. A. Nishioka, Toyo Oil Manufacturing Co., Chiba, Japan. **Uses.** The seeds are widely used for the production of rape-seed oil. **Previous work on 4,4-dimethylsterols (triterpene alcohols).** Cycloartenol [1], cycloartenol [1–4], 24-methylenecycloartenol [1–4],  $\alpha$ -amyrin [2,3],  $\beta$ -amyrin [1–3], butyrospermol [3,4] and lupeol [3] in the commercially available rape-seed oil produced from the seeds of *Brassica* spp.

**Present work.** Extraction of the dried and ground seeds of *B. napus* (158 g) by  $\text{CH}_2\text{Cl}_2$  using a Soxhlet apparatus gave a dark brown oil (64 g). The unsaponifiable fraction (1.22 g) extracted from the oil after saponification was fractionated by PLC on Si gel using hexane– $\text{Et}_2\text{O}$  (4:1) as the developing solvent [1], to give the 4,4-dimethylsterol fraction (36 mg). On acetylation, the acetate fraction (33 mg) obtained was separated into two principal bands ( $R_f$  0.39 and 0.24) by PLC on  $\text{AgNO}_3$ –Si gel (1:9) using  $\text{CH}_2\text{Cl}_2$ – $\text{CCl}_4$  (1:1) as the developing solvent. The acetate fraction (6 mg) recovered from the  $R_f$  0.24 band gave two GLC peaks: The GC–MS for the faster eluted

minor peak (20%) gave  $\text{M}^+$  at  $m/e$  482 ( $\text{C}_{33}\text{H}_{54}\text{O}_2$ ) with fragment ions at  $m/e$  467 (Table 1, ion a) and 407 (c) indicating that this compound was the acetate of a  $\text{C}_{31}$  triterpene alcohol. Ions at  $m/e$  297 (d) and 301 (e) showed that the side chain had nine carbon atoms and one double bond. An ion at  $m/e$  383 (g) indicated the presence of a 24-methylene group [5,6] in sterol devoid of the 9 $\beta$ ,19-cyclopropane ring [6], and the ion e indicated the presence of a methyl group attached at C-14 [7]. The GLC and MS data of this triterpene acetate were identical with those measured for the acetate of 24-methylenelanost-8-en-3 $\beta$ -ol occurring in the acetylated 4,4-dimethylsterol fraction separated from an yeast, *Pichia* sp. [8]. The slower eluted major peak (80%) was identified as the acetate of 24-methylenecycloartenol by comparison of its GLC and MS data with those of the authentic specimen of 24-methylenecycloartenyl acetate [9]. The hydrogenated fraction ( $\text{PtO}_2$  catalyst,  $\text{Et}_2\text{O}$  soln) afforded two constituents which had the GLC and MS data identical with those of the authentic specimens [9] of the acetates of 24-methyllanost-8-en-3 $\beta$ -ol and 24-methylcycloartenol, respectively.

Thus, 24-methylene-5 $\alpha$ -lanost-8-en-3 $\beta$ -ol occurs as a

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Table 1. GLC and MS data of triterpene acetates.

	GLC data				MS data (in GC-MS, 70 eV), <i>m/e</i> (>200)						
	<i>RR</i> <sub>f</sub> (sitosterol = 1.00)				and relative intensities (%)						
	SE-30*	OV-17†	OV-225‡	M <sup>+</sup>	a§	b	c	d	e	f	g
24-Methylenecycloartanyl acetate	1.77	1.67	1.45	482 (16)	467 (14)	422 (100)	407 (84)	297 (34)	301 (14)	300 (36)	—
24-Methylcycloartanyl acetate	1.82	1.62	1.37	484 (18)	469 (24)	424 (100)	409 (90)	297 (59)	—	302 (56)	—
24-Methylenelanost-8-en-3β-yl acetate	1.56	1.41	1.20	482 (39)	467 (100)	—	407 (89)	297 (6)	301 (28)	—	383 (11)
24-Methyllanost-8-en-3β-yl acetate	1.61	1.37	1.13	484 (20)	469 (100)	424 (7)	409 (98)	297 (7)	301 (7)	—	—

\* 3% SE-30; 238°,  $\text{N}_2$  50 ml/min. † 3% OV-17; 268°,  $\text{N}_2$  50 ml/min. ‡ 10% OV-225; 260°,  $\text{N}_2$  50 ml/min. § Major fragment ions: a,  $\text{M}^+$ –Me; b,  $\text{M}^+$ –AcOH; c,  $\text{M}^+$ –Me–AcOH; d,  $\text{M}^+$ –SC–AcOH; e,  $\text{M}^+$ –SC–56; f,  $\text{M}^+$ –ring A [6]; g,  $\text{M}^+$ –Me–84.